Experiments of Physical Chemistry Laboratory

S	Exp. Name	Description	Tools & instruments	Link
1	Refractive Index	Studying how the refractive index	1. Refractometer	https://www.youtu
		of alcohol changes with varying	2. Test Tubes	be.com/watch?v=-
		concentrations.	3. Alcohol Solutions of	<u>txsagjBRqw</u>
			Varying Concentrations	
2	Finding Molecular	Determination the molecular	1. Conical Flask	https://www.youtu
	Weight Using Dumas	weight of a volatile liquid by	2. Hot Plate	be.com/watch?v=0
	Method	measuring the mass of gas	3. Thermometer	UJXa9Hd88I
		produced when the liquid is	4. Balance (Precision)	
		vaporized. and the pressure,	5. Rubber Stopper with	
		temperature, and volume of the gas	Hole	
		are measured. The ideal gas law	6.Sample of Volatile Liquid	
		(PV = nRT) is applied.		
3	Calorimeter	Determine the amount of heat	1. Calorimeter	https://www.youtu
		released when an acid and a base	2. Thermometer	be.com/watch?v=E
		react to form water and a salt. This	3. Balance (Precision)	LSu7zmVRMg
		is typically done by mixing a	4. Burette or Pipette	
		known concentration of acid and	5. Graduated Cylinder	
		base in a calorimeter and measuring	6. Hydrochloric Acid	
		the temperature change during the	7. Sodium Hydroxide	
		reaction.	8. pH Meter or pH	

4	Measuring Viscosity and Relative Viscosity	Measure the viscosity of a liquid relative to the viscosity of a reference liquid (typically water) by using a viscometer.	 Viscometer Thermometer Stopwatch Stirring Rod Sample of Liquid 	$\frac{https://www.youtu}{be.com/watch?v=I0}$ $\frac{aYfmbGmSA\&t=1}{s}$
			6. Ruler or Calipers	
5	Spectrophotometer	Determination the absorbance of a product formed during a chemical reaction as a function of the reactant concentrations using a visible-range spectrophotometer.	 Visible Range Spectrophotometer Cuvettes Pipettes and Pipette Tips Volumetric Flasks Test Tubes or Beakers Samples of Reactants 	https://www.youtu be.com/watch?v=v JCnW1-p_d0